

AGELASPHINS, NOVEL α -GALACTOSYLCERAMIDES FROM THE MARINE SPONGE *AGELAS MAURITIANUS*

Takenori Natori,^{a*} Yasuhiko Koezuka,^a and Tatsuo Higa^b

^a Pharmaceutical Research Laboratory, Kirin Brewery Co., Ltd., 3 Miyahara-cho, Takasaki, Gunma 370-12, Japan

^b Department of Marine Sciences, University of the Ryukyus, Nishihara, Okinawa 903-01, Japan

Abstract: New glycosphingolipids, agelasphins (1-4), have been isolated from the marine sponge *Agelas mauritianus*, and their structures were determined. Agelasphins having antitumor activity are the first examples of galactosylceramides containing an α -galactosyl linkage.

In our quest for bioactive substances from marine organisms, we found novel glycosphingolipids possessing an α -galactosylceramide structure from an extract of the sponge *Agelas mauritianus* Carter (1883).¹ To the best of our knowledge, this is the first report on the isolation of α -galactosylceramides from natural sources. We herein describe the isolation and the structure elucidation of these novel substances, named agelasphins (AGLs).

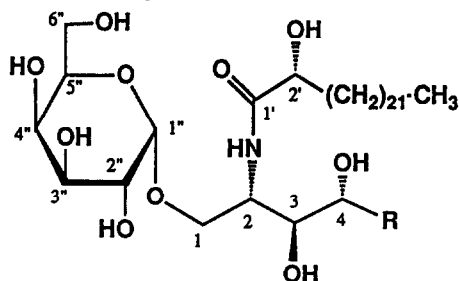
A lyophilized sample of the sponge was extracted with 1:1 CHCl₃-MeOH. The residue of the extract was partitioned between EtOAc and water. The EtOAc layer was washed with 30% aq. MeOH to remove polar constituents containing agelasines.² The remaining lipophilic portion was repeatedly chromatographed on silica gel to give a fraction which was shown to be virtually a mixture of homologous glycosphingolipids as demonstrated by a series of molecular ion peaks at *m/z* 816, 830, 844, 858, and 872 in the negative FABMS. The IR spectrum of the fraction exhibited strong absorption bands for hydroxyl and amide groups. The ¹³C NMR spectrum (C₅D₅N) showed, in addition to the amide functionality (δ 175.0, 50.6), the presence of an anomeric carbon (δ 101.2) and a number of aliphatic carbons, suggesting the major portion of the fraction being cerebrosides. The presence of galactose as the common sugar moiety was shown by methanolysis of the fraction followed by acetylation and analysis of the resulting sugar.³ Optical rotation ($[\alpha]_D^{23} +142.3^\circ \rightarrow +78.2^\circ$, H₂O, 24 h) of the free sugar revealed that it was D-galactose. Furthermore, a signal at δ 5.58 (d, *J* = 3.7Hz) in the ¹H NMR spectrum of the intact fraction indicated the stereochemistry of the anomeric position to be α .

The fraction could be separated using reversed-phase HPLC (ODS, MeOH) to give six major constituents designated AGL-7a (1), -7b, -9a (2), -9b (3), -11 (4), and -13 in 0.0022, 0.0027, 0.0019, 0.0080, 0.0096, and 0.0009% yield, respectively, along with some minor ones. The structures of four (1-4) of the AGLs are reported in this paper. All the compounds gave essentially the same IR absorptions (3400, 2950, 2870, 1645, 1535, 1475, and 1080 cm⁻¹) and ¹H and ¹³C NMR data except for minor differences.⁴ Thus, structure elucidation is described for agelasphin-9b (3) as a representative.

Agelasphin-9b (3) was analyzed for C₄₈H₉₄NO₁₀ [(M-H)⁻ *m/z* 844.6948, Δ 6.5mmu] by negative HR FABMS. In the ¹H NMR spectrum in C₅D₅N 3 showed eight exchangeable proton signals due to one NH [δ 8.50 (d, *J* = 9.2 Hz)] and seven OH (broad singlets or doublets in the range of δ 6 - 7.5). The anomeric proton was observed as a doublet (*J* = 3.7 Hz) at δ 5.59. The coupling constant clearly indicated its β -orientation in the galactose moiety. The NMR data for the core portion of the molecule were assigned by 2D NMR experiments.⁴ In methanolysis with aq. HCl-MeOH⁵ 3 gave 1-*O*-methylgalactose, methyl 2-(*R*)-hydroxytetraacosanoate (5), and a long-chain base. The structure of the ester 5 which could be separated from the reaction mixture by extraction with hexane was identified by spectroscopic data⁶ and confirmed including the stereochemistry by synthesis.⁷ The mixture of base and sugar was separated on an

Amberlite CG-400 column by eluting with MeOH. The base (phytosphingosine) was acetylated and identified by spectroscopic data to be 2-acetoamino-1,3,4-tri-*O*-acetyl-16-methyl-1,3,4-hexadecanetriol (6).⁸ The stereochemistry of 6 was also established by synthesis.⁷ The overall structure of 3 was unambiguously established by a total synthesis.⁷ The structures for the remaining AGLs were determined by spectroscopic comparison with 3 and methanolytic analysis.

Although a number of β -galactosylceramides have been known, AGLs are the first cerebrosides having an α -galactosyl linkage. All AGLs showed antitumor activity.⁹



	R
Agelasphin-7a (1)	$-(\text{CH}_2)_{11}\text{-CH}_3$
Agelasphin-9a (2)	$-(\text{CH}_2)_{12}\text{-CH}_3$
Agelasphin-9b (3)	$-(\text{CH}_2)_{11}\text{-CH}(\text{CH}_3)_2$
Agelasphin-11 (4)	$-(\text{CH}_2)_{11}\text{-CH}(\text{CH}_3)\text{-C}_2\text{H}_5$

References and Notes

- The sponge was collected in Okinawa in 1988 and identified by Dr. Ole Tendal.
- H. Nakamura, H. Wu, Y. Ohizumi, and Y. Hirata, *Tetrahedron Lett.*, 25, 2989 (1984).
- Y. Kawano, R. Higuchi, R. Isobe, and T. Komori, *Liebigs Ann. Chem.*, 19 (1988).
- Full ¹H and ¹³C NMR data are given for 1, while only signals different from 1 are listed for the rest.
AGL-7a (1): $[\alpha]_{\text{D}}^{24} +52.3^\circ$ (*c* 1.0, Pyr); mp 193.5-195.0 °C; HR FABMS *m/z* 816.6619 [(M-H)⁻], calcd. for C₄₆H₉₀NO₁₀ 816.6557; ¹H NMR (500 MHz, C₅D₅N) δ 8.49 (1H, d, *J* = 9.2 Hz, NH), 7.53 (1H, bs, OH), 7.04 (1H, bs, OH), 6.71 (1H, d, *J* = 6.7 Hz, OH), 6.68 (1H, bs, OH), 6.52 (1H, bs, OH), 6.32 (1H, bs, OH), 6.09 (1H, d, *J* = 6.1 Hz, OH), 5.58 (1H, d, *J* = 3.7 Hz, H1'), 5.26 (1H, m, H2), 4.62 (2H, m, H2', H2''), 4.57 (1H, m, H1a), 4.52 (1H, bs, H4''), 4.48 (2H, m, H5', H3''), 4.37 (1H, m, H3), 4.34 (2H, m, H1b, H6'a), 4.32 (1H, m, H6'b), 4.26 (1H, m, H4), 2.28 (1H, m), 2.18 (1H, m), 1.98 (1H, m), 1.87 (2H, m), 1.73 (1H, m), 1.66 (2H, m), 1.10-1.40 (56H, m), and 0.85 (6H, t, *J* = 7.3 Hz); ¹³C NMR (125 MHz, C₅D₅N) δ 175.0 (s, C1'), 101.2 (d, C1''), 76.5 (d, C3), 73.0 (d, C5''), 72.4 (d, C2'), 72.3 (d, C4), 71.6 (d, C3''), 70.9 (d, C4''), 70.1 (d, C2''), 68.1 (t, C1), 62.6 (t, C6''), 50.4 (d, C2), 35.5 (t), 34.4 (t), 32.1 (t), 30.3 (t), 30.1 (t), 30.0 (t), 29.9 (t), 29.5 (t), 26.4 (t), 25.8 (t), 22.9 (t), and 14.2 (q).
AGL-9a (2): $[\alpha]_{\text{D}}^{24} +49.9^\circ$ (*c* 1.0, Pyr); mp 201.0-203.5 °C; HR FABMS *m/z* 830.6747 [(M-H)⁻], calcd. for C₄₇H₉₂NO₁₀ 830.6726; ¹H NMR same to 1 except for the integration for the CH₂ signal (δ 1.1*l*-1.40); ¹³C-NMR same to 1.
AGL-9b (3): $[\alpha]_{\text{D}}^{24} +55.5^\circ$ (*c* 1.0, Pyr); mp 211.0-212.0 °C; HR FABMS 844.6948 [(M-H)⁻], calcd. for C₄₈H₉₄NO₁₀ 844.6883; ¹H NMR δ 1.46 (2H, m), 0.84 (3H, t, *J* = 6.1 Hz), and 0.84 (6H, d, *J* = 6.7 Hz); ¹³C NMR δ 50.5 (d, C2), 39.2 (t), 28.2 (d), 27.7 (t), and 22.7 (q).
AGL-11 (4): $[\alpha]_{\text{D}}^{24} +51.9^\circ$ (*c* 1.0, Pyr); mp 189.5-190.5 °C; HR FABMS 858.7032 [(M-H)⁻], calcd. for C₄₉H₉₆NO₁₀ 858.7040; ¹H NMR δ 0.88 (9H, m); ¹³C NMR (55 °C) δ (ppm) 176.4 (s, C1'), 51.5 (d, C2), 37.4 (t), 35.1 (d), 33.6 (t), 19.9 (q), and 12.0 (q).
- R. C. Gaver and C. C. Sweeley, *J. Am. Oil Chem. Soc.*, 42, 294 (1965). R. Higuchi, T. Natori and T. Komori, *Liebigs Ann. Chem.*, 51 (1990).
- 5: $[\alpha]_{\text{D}}^{23} -2.4^\circ$ (*c* 3.0, CHCl₃); FDMS *m/z* 398 (M⁺); ¹H NMR (500 MHz, CDCl₃) δ 4.19 (1H, dd, *J* = 4.3, 7.3 Hz, H2), 3.79 (3H, s, OCH₃), 2.74 (1H, d, OH), 1.78 (1H, m), 1.64 (1H, m), 1.18-1.50 (40H, m), and 0.88 (3H, t, *J* = 7.3 Hz, CH₃).
- See accompanying paper.
- 6: $[\alpha]_{\text{D}}^{23} +23.4^\circ$ (*c* 0.3, CHCl₃); FDMS *m/z* 486 [(M+H)⁺]; ¹H NMR (500 MHz, CDCl₃) δ 5.97 (1H, d, *J* = 9.2 Hz, NH), 5.10 (1H, dd, *J* = 8.5, 3.1 Hz, H3), 4.93 (1H, dt, *J* = 9.8, 3.1 Hz, H4), 4.47 (1H, m, H2), 4.29 (1H, dd, *J* = 11.6, 4.3 Hz, H1a), 4.00 (1H, dd, *J* = 11.6, 3.1 Hz, H1b), 2.08 (3H, s, 3-OAc), 2.05 (6H, s, 1-OAc, 4-OAc), 2.03 (3H, s, NHAc), 1.12-1.70 (23H, m), and 0.86 (6H, d, *J* = 6.7 Hz).
- Antitumor activity will be reported elsewhere.

(Received in Japan 3 February 1993; accepted 28 April 1993)